



# The periodic table

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1 Hydrogen <b>H</b>	2 Helium <b>He</b>	3 Lithium <b>Li</b>	4 Beryllium <b>Be</b>	5 Boron <b>B</b>	6 Carbon <b>C</b>	7 Nitrogen <b>N</b>	8 Oxygen <b>O</b>	9 Fluorine <b>F</b>	10 Neon <b>Ne</b>	11 Sodium <b>Na</b>	12 Magnesium <b>Mg</b>	13 Aluminum <b>Al</b>	14 Silicon <b>Si</b>	15 Phosphorus <b>P</b>	16 Sulfur <b>S</b>	17 Chlorine <b>Cl</b>	18 Argon <b>Ar</b>	
19 Potassium <b>K</b>	20 Calcium <b>Ca</b>	21 Scandium <b>Sc</b>	22 Titanium <b>Ti</b>	23 Vanadium <b>V</b>	24 Chromium <b>Cr</b>	25 Manganese <b>Mn</b>	26 Iron <b>Fe</b>	27 Cobalt <b>Co</b>	28 Nickel <b>Ni</b>	29 Copper <b>Cu</b>	30 Zinc <b>Zn</b>	31 Gallium <b>Ga</b>	32 Germanium <b>Ge</b>	33 Arsenic <b>As</b>	34 Selenium <b>Se</b>	35 Bromine <b>Br</b>	36 Krypton <b>Kr</b>	
37 Rubidium <b>Rb</b>	38 Strontium <b>Sr</b>	39 Yttrium <b>Y</b>	40 Zirconium <b>Zr</b>	41 Niobium <b>Nb</b>	42 Molybdenum <b>Mo</b>	43 Technetium <b>Tc</b>	44 Ruthenium <b>Ru</b>	45 Rhodium <b>Rh</b>	46 Palladium <b>Pd</b>	47 Silver <b>Ag</b>	48 Cadmium <b>Cd</b>	49 Indium <b>In</b>	50 Tin <b>Sn</b>	51 Antimony <b>Sb</b>	52 Tellurium <b>Te</b>	53 Iodine <b>I</b>	54 Xenon <b>Xe</b>	
55 Cesium <b>Cs</b>	56 Barium <b>Ba</b>	57-70 Lanthanum <b>La</b>	71 Lutetium <b>Lu</b>	72 Hafnium <b>Hf</b>	73 Tantalum <b>Ta</b>	74 Tungsten <b>W</b>	75 Rhenium <b>Re</b>	76 Osmium <b>Os</b>	77 Iridium <b>Ir</b>	78 Platinum <b>Pt</b>	79 Gold <b>Au</b>	80 Mercury <b>Hg</b>	81 Thallium <b>Tl</b>	82 Lead <b>Pb</b>	83 Bismuth <b>Bi</b>	84 Polonium <b>Po</b>	85 Astatine <b>At</b>	86 Radon <b>Rn</b>
87 Francium <b>Fr</b>	88 Radium <b>Ra</b>	89-102 Actinium <b>Ac</b>	103 Lawrencium <b>Lr</b>	104 Rutherfordium <b>Rf</b>	105 Dubnium <b>Db</b>	106 Seaborgium <b>Sg</b>	107 Bohrium <b>Bh</b>	108 Hassium <b>Hs</b>	109 Meitnerium <b>Mt</b>	110 Darmstadtium <b>Ds</b>	111 Roentgenium <b>Rg</b>	112 Copernicium <b>Cn</b>	113 Ununtrium <b>Uut</b>	114 Flerovium <b>Fl</b>	115 Ununpentium <b>Uup</b>	116 Livermorium <b>Lv</b>	117 Ununseptium <b>Uus</b>	118 Ununoctium <b>Uuo</b>

Key:  
Element Name  
Atomic number  
Symbol  
Atomic weight (mean relative mass)

Lanthanum 57 <b>La</b>	Cerium 58 <b>Ce</b>	Praseodymium 59 <b>Pr</b>	Neodymium 60 <b>Nd</b>	Promethium 61 <b>Pm</b>	Samarium 62 <b>Sm</b>	Europium 63 <b>Eu</b>	Gadolinium 64 <b>Gd</b>	Terbium 65 <b>Tb</b>	Dysprosium 66 <b>Dy</b>	Holmium 67 <b>Ho</b>	Erbium 68 <b>Er</b>	Thulium 69 <b>Tm</b>	Ytterbium 70 <b>Yb</b>
Actinium 89 <b>Ac</b>	Thorium 90 <b>Th</b>	Protactinium 91 <b>Pa</b>	Uranium 92 <b>U</b>	Neptunium 93 <b>Np</b>	Plutonium 94 <b>Pu</b>	Americium 95 <b>Am</b>	Curium 96 <b>Cm</b>	Berkelium 97 <b>Bk</b>	Californium 98 <b>Cf</b>	Einsteinium 99 <b>Es</b>	Fermium 100 <b>Fm</b>	Mendelevium 101 <b>Md</b>	Nobelium 102 <b>No</b>

\*lanthanoids

\*\*actinoids

**Symbols and names:** the symbols and names of the elements, and their spellings are those recommended by the International Union of Pure and Applied Chemistry (IUPAC - <http://www.iupac.org/>). Names have yet to be proposed for elements 113, 115, 117, and 118 and so those used here are IUPAC's temporary systematic names. In some countries, the spellings **aluminium**, **cesium**, and **sulphur** are usual.

**Group labels:** the numeric system (1–18) used here is the current IUPAC convention.

**Atomic weights (mean relative masses):** these are the IUPAC 2009 values and given to 5 significant figures. The last significant figure of each value is considered reliable to ±1 except where a larger uncertainty is given in parentheses. Representative values for those elements having an atomic weight interval are given (H, Li, B, C, N, O, Si, S, Cl, Ti). Elements for which the atomic weight is given within brackets have no stable nuclides and are represented by the element's longest lived isotope reported in the IUPAC 2009 values.

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