Answer Key for Isotope Cards

| | |
|---|---|
|  | Alpha<br>Element has more than 82 protons |
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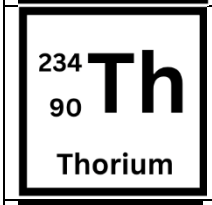
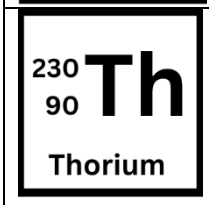
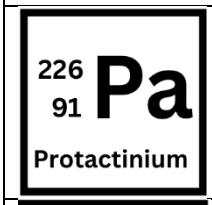
|  | Alpha<br>Element has more than 82 protons |
|  | Alpha<br>Element has more than 82 protons |
|  | Alpha<br>Element has more than 82 protons |
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|---|---|
| $^{218}_{84}Po$ Polonium | Alpha<br>Element has more than 82 protons |
| $^{211}_{83}Bi$ Bismuth | Alpha<br>Element has more than 82 protons |
| $^{210}_{82}Pb$ Lead | Beta -<br>There are too many neutrons for the number of protons |
| $^{138}_{56}Ba$ Barium | Stable |
| $^{125}_{55}Cs$ Caesium | Beta +<br>There are not enough neutrons for the number of protons |
| $^{58}_{30}Zn$ Zinc | Beta +<br>There are not enough neutrons for the number of protons |
| $^{12}_{6}C$ Carbon | Stable |
| $^{22}_{9}F$ Fluorine | Beta -<br>There are too many neutrons for the number of protons |

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|---|---|
| $^{184}_{74}W$ Tungsten | Stable |
| $^{75}_{33}As$ Arsenic | Stable |
| $^{146}_{55}Cs$ Caesium | Beta -<br>There are too many neutrons for the number of protons |
| $^{73}_{30}Zn$ Zinc | Beta -<br>There are too many neutrons for the number of protons |
| $^{16}_{6}C$ Carbon | Beta -<br>There are too many neutrons for the number of protons |
| $^{19}_{9}F$ Fluorine | Stable |
| $^{6}_{3}Li$ Lithium | Stable |
| $^{56}_{26}Fe$ Iron | Stable |

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|---|---|
| $^{86}_{35}Br$ Bromine | Beta -<br>More neutrons than protons |
| $^{114}_{47}Ag$ Silver | Beta -<br>More neutrons than protons |
| $^{150}_{79}Au$ Gold | Beta +<br>More protons than neutrons for the 1:1.5 ratio |
| $^{196}_{77}Ir$ Iridium | Beta -<br>More neutrons than protons |
| $^{10}_{6}C$ Carbon | Beta +<br>More protons than the 1:1 ratio |
| $^{16}_{9}F$ Fluorine | Beta +<br>More protons than the 1:1 ratio |
| $^{32}_{18}Ar$ Argon | Beta +<br>More protons than the 1:1 ratio |
| $^{57}_{27}Co$ Cobalt | Beta +<br>Slightly more protons than the 1:1/1:1.5 ratios that the light near the pattern of. |
| $^{122}_{53}I$ Iodine | Beta - |

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|---|---|
| $^{125}_{49}In$<br>Indium | Beta -<br>More than 1.5 neutrons than protons |
| $^{194}_{80}Hg$<br>Mercury | Beta+<br>More protons than the 1:1.5 p+/n0 ratio. neutrons |
| $^{210}_{83}Pb$<br>Lead | Alpha<br>Element has more than 82 protons |
| $^{167}_{65}Tb$<br>Terbium | Beta -<br>More neutrons than protons |
| $^{128}_{50}Sn$<br>Tin | Beta -<br>More neutrons than protons |
| $^{67}_{25}Mn$<br>Manganese | Beta -<br>More neutrons than protons |
| $^{44}_{16}S$<br>Sulfur | Beta -<br>More neutrons than protons |

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